



**DP-003-1104016**

Seat No. \_\_\_\_\_

**M. Sc. (Sem. IV) Examination**

**March - 2022**

**C(I)-404 : Inorganic Chemistry**

*(Coordination Chemistry)*

**Faculty Code : 003**

**Subject Code : 1104016**

Time :  $2\frac{1}{2}$  Hours]

[Total Marks : 70

- Instructions :** (1) Answer all questions.  
(2) All questions carry equal marks.

- 1** Answer the following (Any Seven) **14**
- (a) Give the principle of Mole ratio method.
  - (b) Give the reaction pathway for conversion of trans-[Co(en)<sub>2</sub>Cl<sub>2</sub>] to cis-[Co(en)<sub>2</sub>Cl<sub>2</sub>].
  - (c) Explain isomerism reactions of aquo and hydroxo complexes of cobalt.
  - (d) Discuss complementary two electron transfer reaction with example.
  - (e) Define formation function and fraction of complex formation.
  - (f) Discuss theories of trans effect.
  - (g) Give principle of jobs method.
  - (h) What do you mean by stability of complex?
  - (i) Give Associative mechanism for nucleophilic substitution reaction in octahedral metal complexes.
  - (j) Name different methods to determine stability constant.
- 2** Answer the following : (Any Two) **14**
- (a) Explain Correction method.
  - (b) Discuss Vosburgh and Copper correction to the Job's method.
  - (c) What is stability? Define step wise and overall stability constant and obtain the reaction between them.

- 3** Answer the following : (Any Two) **14**
- (a) Discuss half integral method to obtain the stepwise stability constant for  $ML_2$  system.
  - (b) Explain Molecular Rearrangement process proceeds by  $SN^1$  mechanism.
  - (c) Give an account of ligand substitution reaction with suitable example.
  - (d) Explain the trans effect with suitable example. Explain the outer sphere electron transfer reaction in coordination compound.
- 4** Answer the following : **14**
- (a) Discuss job's method of continuous variation when more than one complex is present in solution.
  - (b) Explain Bjerrum's-formation-functions.
- 5** Explain the replacement mechanism of coordinate water in octahedral complex. **14**

**OR**

- 5** Discuss the Laboratory method for pH titration technique to find out the Stepwise stability constant. **14**

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